



# Friday 10 June 2016 - Morning

# GCSE GATEWAY SCIENCE SCIENCE B

**B712/02** Science modules B2, C2, P2 (Higher Tier)

Candidates answer on the Question Paper. A calculator may be used for this paper.

OCR supplied materials:

None

Other materials required:

- Pencil
- Ruler (cm/mm)

**Duration:** 1 hour 30 minutes



Candidate forename				Candidate surname			
Centre numb	oer			Candidate nu	umber		

### **INSTRUCTIONS TO CANDIDATES**

- Write your name, centre number and candidate number in the boxes above. Please write clearly and in capital letters.
- Use black ink. HB pencil may be used for graphs and diagrams only.
- Answer all the questions.
- Read each question carefully. Make sure you know what you have to do before starting your answer.
- Write your answer to each question in the space provided. If additional space is required, you should use the lined page(s) at the end of this booklet. The question number(s) must be clearly shown.
- Do not write in the bar codes.

# **INFORMATION FOR CANDIDATES**

- The quality of written communication is assessed in questions marked with a pencil ( ).
- A list of equations can be found on page 2.
- The Periodic Table can be found on the back page.
- The number of marks is given in brackets [ ] at the end of each question or part question.
- The total number of marks for this paper is 85.
- This document consists of 28 pages. Any blank pages are indicated.



# **EQUATIONS**

energy = mass x specific heat capacity x temperature change
energy = mass x specific latent heat

efficiency = 
$$\frac{\text{useful energy output (x 100\%)}}{\text{total energy input}}$$

wave speed = frequency × wavelength

power = voltage × current

energy supplied = power x time

average speed = 
$$\frac{\text{distance}}{\text{time}}$$

distance = average speed x time

$$s = \frac{(u+v)}{2} \times t$$

$$acceleration = \frac{change in speed}{time taken}$$

force = mass x acceleration

weight = mass x gravitational field strength

work done = force  $\times$  distance

$$power = \frac{work done}{time}$$

 $power = force \times speed$ 

$$KE = \frac{1}{2}mv^2$$

momentum = mass x velocity

$$force = \frac{change \ in \ momentum}{time}$$

$$GPE = mgh$$

$$mgh = \frac{1}{2}mv^2$$

$$resistance = \frac{voltage}{current}$$

# Answer **all** the questions.

# **SECTION A – Module B2**

1 Look at the picture of a sperm whale.



(a)	Spe	im whales were once nunted.	
	Hur	ting of sperm whales was made illegal to prevent the number falling to a critical level.	
	(i)	Explain why it is important that the number of sperm whales does not fall to a critical leve	؛l.
		[1	]
	(ii)	Describe <b>two</b> reasons why it is difficult to prevent whales being hunted.	
		1	
		2	
(h)	Son	[2] ne other species of whales are still being hunted as <b>sustainable resources</b> .	<u>'</u> ]
(D)	3011	le other species of whates are still being hunted as sustainable resources.	
	Ехр	lain how whale populations can be sustained.	

[Total: 5]

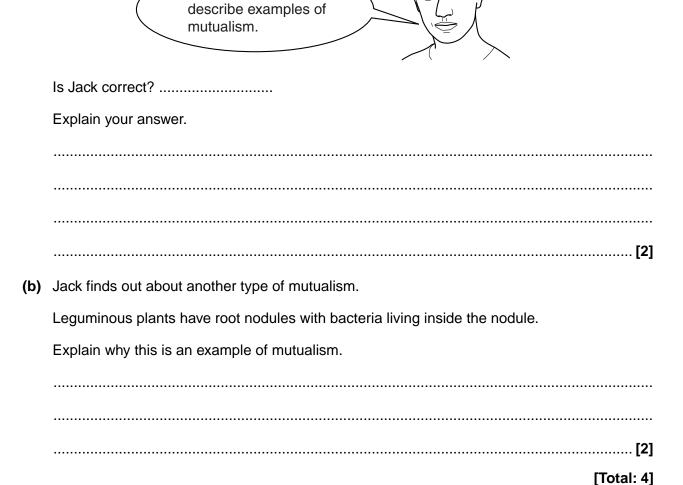
Turn over

2 Jack reads this information about relationships between organisms.

Organisms involved	Information about the relationship between the organisms				
bees	Rees transfer pollen from one flower to another				
and	Bees transfer pollen from one flower to another.				
flowers	Bees make less honey in years when there are fewer flowers.				
acacia trees	Ants feed on nectar from the trees.				
and					
ants	The ants kill trees next to the acacia tree they live on.				
mistletoe plants	The mistletoe plant gets nutrients from the tree.				
and	Trees with mistletoe growing on them are usually smaller than those with no mistletoe.				
birch trees					

(a) Jack makes a statement about the relationships.

All of the relationships



3 Look at the picture of some limestone rock.



(a) The limestone formed millions of years ago at the bottom of a shallow sea.

Acid rain is now weathering the rock.

Explain how carbon became locked up in the limestone and how weathering contributes to global warming.

^
The quality of written communication will be assessed in your answer to this question.
[6]
Acid rain is one type of air pollution.
Some organisms living on trees and rocks can be used to indicate the levels of air pollution.
These organisms are one of many different types of indicator species.
Write down the name of this type of indicator species that lives on trees and rocks.
[1]

[Total: 7]

**Turn over** 

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(b)

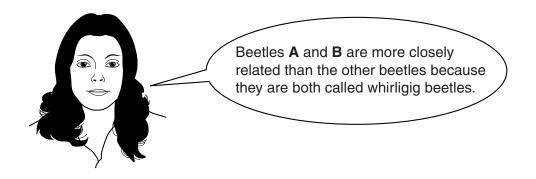
4 Matt and Ellie are investigating the animals living in a lake.

They collect some of the animals.

(a) Ellie identifies some of the beetles they collect and writes their names in a table.

Beetle	Common name Binomial na		
Α	whirligig beetle Gyrinus natator		
В	hairy whirligig beetle Orectochilus villosu		
С	great diving beetle	Dytiscus marginalis	
D	there is no common name for this beetle	Dytiscus latissimus	

Ellie makes this statement about the beetles.

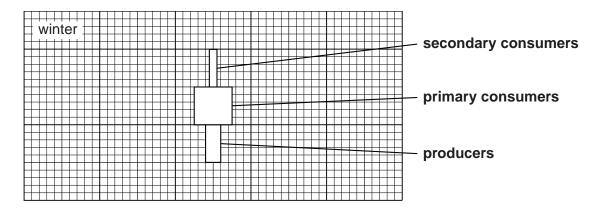


[	1
Explain your answer.	
Is Ellie correct?	

(b) Matt finds some information about the biomass values for the lake in winter and spring.
Look at the table.

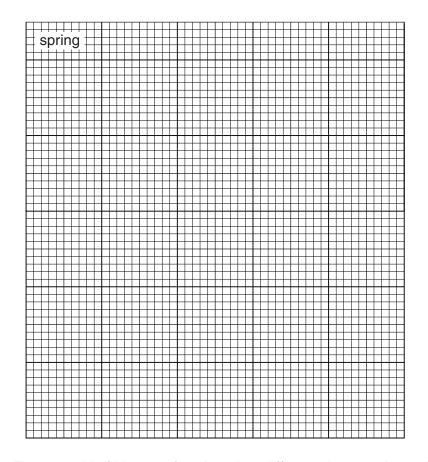
Trophia lovel	Biomass in mg dry mass per m <sup>3</sup>			
Trophic level	Winter	Spring		
producers	4	100		
primary consumers	10	12		
secondary consumers	2	6		

The diagram below shows the pyramid of biomass for winter.



Draw the pyramid of biomass for spring on the grid below.

Make sure the bars are drawn to the same scale and labelled.



[2]

The pyramid of biomass for winter is a different shape to the spring pyramid of biomass.

Describe one way the winter pyramid is different in shape and suggest a reason for the difference.

[Total: 5] Turn over

**5** Look at the pictures of different birds.







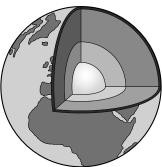
herring gull penguin

(a)	Classifying these birds as seabirds is an <b>artificial</b> classification.
	Explain why.
	[1]
(b)	Puffins live in the Arctic.
	Penguins live in the Antarctic.
	Suggest reasons why puffins and penguins are both similar and different.
	[3]
	[Total: 4]

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# 9 SECTION B - Module C2

6 Look at the diagram of the structure of the Earth.



(a)	The surface of the Earth is made up of tectonic plates.
	This was first suggested by a scientist called Wegener in 1914.
	The theory of tectonic plates is now widely accepted.
	Explain why theories proposed by scientists like Wegener take some time to become widely accepted.
	[2]
(b)	The theory of plate tectonics can be used to explain <b>subduction</b> .
	What is subduction and how does it happen?
	You may wish to draw a <b>labelled</b> diagram.

[Total: 5]

Turn over

7 Rachel is making some fertilisers by neutralising acids with alkalis.

Complete the table.

Name of fertiliser	Name of alkali used	Name of acid used	
ammonium phosphate	ammonia	phosphoric acid	
potassium nitrate	potassium hydroxide		
	ammonia	sulfuric acid	

[2]

[Total: 2]

# 11 BLANK PAGE

Question 8 begins on page 12

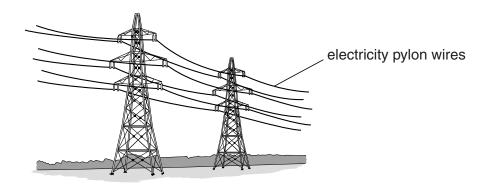
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**8** Look at the table. It shows some properties of different metals.

Metal	Density in g/cm <sup>3</sup>	Relative electrical conductivity	Relative strength	Melting point in °C	Cost per tonne in £
Α	8.9	64	13	1083	3800
В	7.7	11	21	1510	440
С	2.7	40	15	660	1350

(a) Look at the diagram. It shows electricity pylon wires.

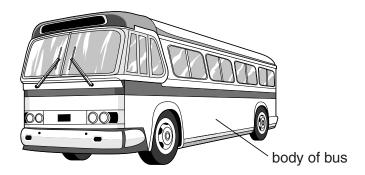


Jo says that metal  $\boldsymbol{\mathsf{A}}$  is the best for making electricity pylon wires.

Dave says that metal **C** is the best.

ra
Evaluate metals <b>A</b> and <b>C</b> for making electricity pylon wires.

(b) Metals A, B and C can be used to make the body of a bus.



escribe the advantages and disadvantages of metals <b>A</b> , <b>B</b> and <b>C</b> for making the body us.	
[Tota]	l: 61

Question 9 begins on page 14

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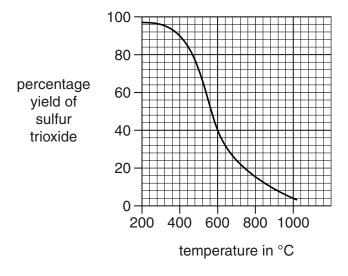
**9** Sulfur trioxide, SO<sub>3</sub>, is made in a chemical factory.

It is used to make sulfuric acid.

The equations show how sulfur trioxide is made.

Look at the graph below.

It shows how the percentage yield of sulfur trioxide changes as the temperature changes.



(a)	How does <b>increasing</b> the temperature affect the percentage yield?
	[1]

[Total: 6]

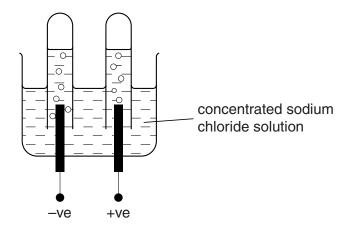
		15		
(b)	The conditions used in the reaction are			
	•	450 °C low pressure catalyst of vanadium(V) oxide.		
	(i)	Suggest why a catalyst is used.		
		[1]		
	(ii) Suggest why a temperature of 450 °C rather than 200 °C is used.			
	Use ideas about rate of reaction and percentage yield.			
		[2]		
(c)	In a	nother industrial process, nitrogen, $N_2$ , reacts with hydrogen, $H_2$ , to make ammonia, $NH_3$ .		
	Writ	te a <b>balanced symbol</b> equation for this reaction.		

Turn over © OCR 2016

**10** Sodium chloride (salt) is a very important chemical.

Concentrated sodium chloride solution can be electrolysed to make useful products.

Look at the diagram below. It shows how this can be done in the laboratory.



Sodium chloride solution contains the ions Na $^+$ , C $l^-$ , H $^+$  and OH $^-$ .

What happens during the electrolysis of concentrated sodium chloride solution?

Your answer should include equations for the reactions at each electrode.

Use e<sup>-</sup> to represent an electron.

The quality of written communication will be assessed in your answer to this question.
 [6]

[Total: 6]

# 17 SECTION C - Module P2

- 11 Nuclear radiation can be alpha, beta or gamma radiation.
  - (a) Draw one line from each radiation to its correct use.

		radiation	use
		alpha	paper thickness gauge
		beta	non-destructive testing
		gamma	smoke detector
			[2
(b)	Polly and	Oliver were talk	ring about the factory that is near their home.
	Polly said	that the factory	produces dangerous radioactive waste.
	Oliver said	that if the was	te was put into thick aluminium cans it could be stored safely.
	Is Oliver c	orrect?	
	Explain yo	our answer.	
			[2
			ITotal: 4

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[Total: 5]

**12** (a) Most of our electrical energy is produced in large power stations.

A large power station produces up to 2000 MW of power.

Some of our electrical energy is now produced by wind farms.

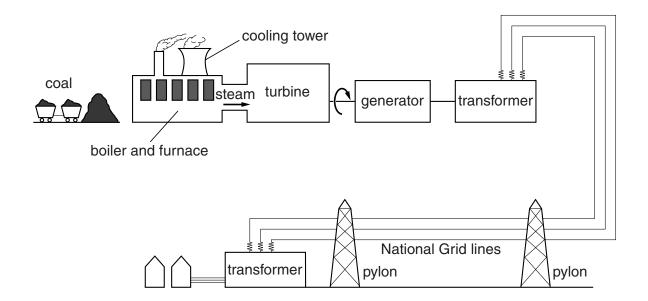


A wind farm is made up of several wind turbines.

Each turbine produces up to 2MW of power.

	Suggest one reason why some people are <b>for</b> more wind farms and one reason why others are <b>against</b> .
	reason for
	reason against
	[2]
(b)	Some companies are now designing and heating their offices using passive solar heating.
	Explain how passive solar heating works and suggest why the position of the windows is important.
	[3]

13 Power stations generate electricity.



Look at the data for a coal-fired power station.

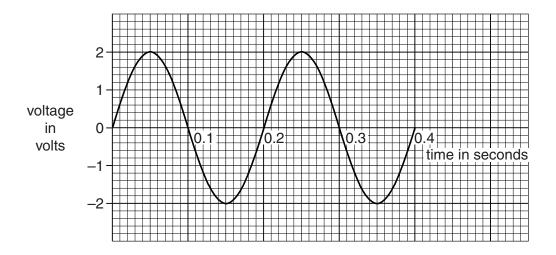
efficiency of power station	30%
average electrical energy generated each second	1.5 × 10 <sup>6</sup> J
heat energy produced by 1 kg of coal	2 × 10 <sup>4</sup> J

(a) Describe in detail the different stages in the production of electricity from this power station

and use the information to calculate the mass of coal (in kg) burnt each second.		
The quality of written communication will be assessed in your answer to this question.		

**(b)** Ruth investigates the output of a model a.c. generator.

Look at the graph below showing the voltage output of the generator.



She increases the number of rotations of the coil each second.

How will this affect the output of this generator?

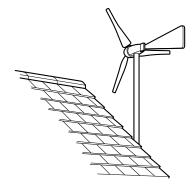
**Draw** your answer on the graph.

[2]

[Total: 8]

14	Sam wants to	reduce his	energy costs.

He fits a wind generator to his house.



The generator produces 1.8 kWh (kilowatt hour) of energy each day (24 hours).

(a) Calculate the average power output of the generator in kW.

	answer	. kW	[2
(b)	Electricity costs 16p per kWh.		
	How much money will Sam save on h	nis electricity bills each day?	
	answer	. pence	[1]

(c) A factory needs 100 kW of electrical power to operate its machinery.

Energy is wasted in transmitting the electricity from the power station to the factory.

Look at the data showing the effect of using different transmission voltages for the power cables to the factory.

Transmission voltage in volts	1000	2000	5000	10 000
Energy wasted per second for each km of cable in W/km	8000	2000	320	80

10000 volts is the most efficient voltage for the transmission of 100kW of electricity.

Explain why.	
	[2]

[Total: 5]

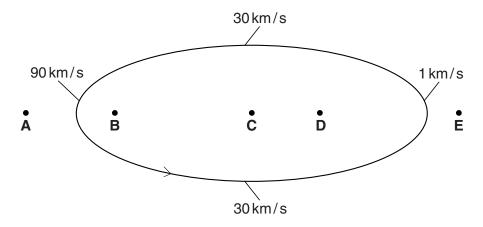
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[Total: 3]

15 Comets are objects in our solar system that orbit the Sun with highly elliptical orbits.

Look at the diagram below showing the path of a comet and its speed at different points on the path.



(a)	Where is the Su	ın on the	diagram	า?						
	Choose from:	Α	В	С	D	E				
	answer									
	Use information	from the	e diagrar	n and ide	eas abou	t forces to	explain y	our answ	er.	
			• • • • • • • • • • • • • • • • • • • •							
										[2]
(b)	There is an aste	eroid belt	betwee	n Mars a	ind Jupite	er.				
	Some people th	ink that	some as	teroids ir	n this bel	t could joir	n togethe	r to form a	new plan	et.
	There is no sign	of this h	nappenin	ıg.						
	Explain why.									

# 23 BLANK PAGE

Question 16 begins on page 24

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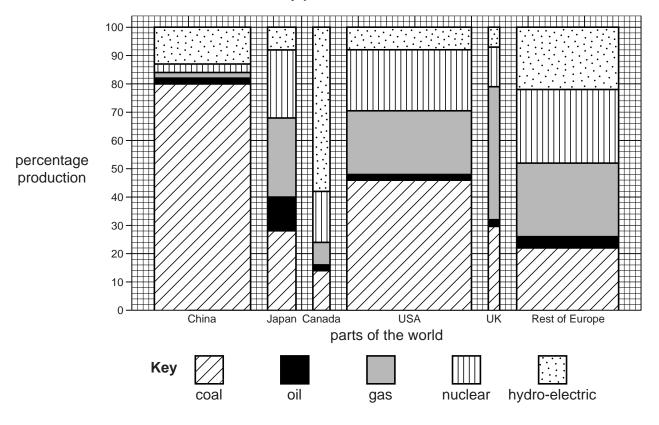
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# 24 SECTION D

**16 (a)** Look at **Graph 1**. This bar chart shows how electricity was produced in different parts of the world.

The width of each bar is a measure of the total amount of electricity produced in 2008.

# **Electricity production in 2008**



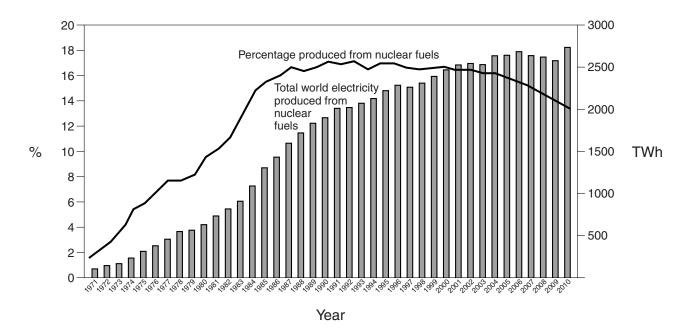
Graph 1

(i)	Look at the percentage of electricity produced from <b>coal</b> in each part of the world.						
	Put these parts of the world in	the correct order. Put the highest first.					
	highest percentage from coal						
	lowest percentage from coal	[2]					
(ii)	China, USA and the rest of Eur	ope generated the largest amounts of electricity in 2008.					
	Suggest why they need to prod	uce the largest amounts of electricity.					
		[1]					
(iii)	What <b>other</b> conclusions can you different parts of the world?	ou make from this bar chart about electricity production in					
		[3]					

**(b)** The production of electricity from nuclear fuels changed between 1971 and 2010.

# Look at Graph 2.

- The bars show the **total** world electricity produced from nuclear fuels (in TWh).
- The line shows the **percentage** of world electricity produced from nuclear fuels.



Graph 2

	what conclusions can you make from <b>Graph 2</b> ?
	[2]
(c)	Using <b>Graphs 1</b> and <b>2</b> , suggest what problems may arise for electricity production during the next 30 years.
	[2]

[Total: 10]

# **END OF QUESTION PAPER**

# 27 ADDITIONAL ANSWER SPACE

If additional space is required, you should use the following lined page(s). The question number(s) must be clearly shown in the margins.



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# The Periodic Table of the Elements

1   2   2   2   2   2   2   2   2   2								
24  86  10  11  11  11  12  14  15  14  15  15  15  16  17  18  18  18  18  18  18  18  18  18	0	4 He	20 <b>Ne</b> neon 10	40 <b>Ar</b> argon 18	84 Kr krypton 36	131 Xe xenon 54	[222] <b>Rn</b> radon 86	t fully
24  86  10  11  11  11  12  14  15  14  15  15  15  16  17  18  18  18  18  18  18  18  18  18	7		19 F fluorine 9	35.5 Ct chlorine 17	80 Br bromine 35	127 <b>I</b> iodine 53	[210] At astatine 85	orted but no
24  86  10  11  11  11  12  14  15  14  15  15  15  16  17  18  18  18  18  18  18  18  18  18	9		16 0 0 0 8	32 S sulfur 16	79 Se selenium 34	128 Te tellurium 52	Po polonium 84	ve been repo
24  86  10  11  11  11  12  14  15  14  15  15  15  16  17  18  18  18  18  18  18  18  18  18	2		14 N nitrogen 7	31 P phosphorus 15		122 Sb antimony 51	209 Bi bismuth 83	i 112-116 hav uthenticatec
1	4		12 C carbon 6	28 Si silicon	73 Ge germanium 32	119 Sn tin 50	207 <b>Pb</b> lead 82	mic numbers a
1	3		11 <b>B</b> boron 5	27 At aluminium 13		115 In indium 49	204 T <b>t</b> thallium 81	nts with atoi
1		·			65 <b>Zn</b> zinc 30	112 Cd cadmium 48	201 <b>Hg</b> mercury 80	Eleme
1					63.5 Cu copper 29	108 <b>Ag</b> silver 47	197 <b>Au</b> gold 79	Rg roentgenium
1					59 Ni nickel 28	106 Pd palladium 46	195 Pt platinum 78	Ds damstadtium
Secondary   Parity   Parity					59 Co cobalt 27	103 Rh rhodium 45	192 <b>Ir</b> iridium 77	[268] Mt meitnerium 109
Secondarium		1 H hydrogen 1			56 Fe iron 26	101 Ru ruthenium 44	190 Os osmium 76	[277] Hs hassium 108
Secondary   Partition   Part					55 Mn manganese 25		186 Re rhenium 75	[264] Bh bohrium 107
9 Be beryllium 4 A 24 Ng magnesium 12 A0 45 Ca Sc Calcium 20 88 89 Sr Y Y Strontium yitrium 2 38 89 Ba La* barium lanthanum 56 56 57 Ra Ac* radium actinium radium 88 Ra Ac*			mass ool number		52 Cr chromium 24	96 Mo molybdenum 42	184 W tungsten 74	[266] Sg seaborgium 106
9 Be beryllium 4 A 24 Ng magnesium 12 A0 45 Ca Sc Calcium 20 88 89 Sr Y Y Strontium yitrium 2 38 89 Ba La* barium lanthanum 56 56 57 Ra Ac* radium actinium radium 88 Ra Ac*		Key	ve atomic omic symk name (proton) r		51 V vanadium 23	93 Nb niobium 41	181 <b>Ta</b> tantalum 73	[262]
2 Be beryillium 4 4 Mg magnesium 12 4 0 Ca calcium 20 88 8 8 Sr strontium 38 Ba bartium 56 Sf Ra radium 888 Ra Ra radium 888 Ra Ra radium 888 Ra Ra radium 888 Ra			relati <b>atc</b> atomic		48 Ti titanium 22	91 Zr zirconium 40	178 Hf hafnium 72	[261] Rf rutherfordium 104
7 9 Li Be lithium 3 4 4 4 4 4 4 4 4 4 6 4 4 6 4 4 6 4 4 6 4 6 4 6 4 4 6 6 4 6 6 4 6 6 4 6					45 Sc scandium 21	89 Y yttrium 39	139 La* lanthanum 57	[227] Ac* actinium 89
7 Li lithium 3 23 Na sodium 11 11 13 Rb rubidium 37 133 Cs caesium 55 Fr franctum 87	2		9 Be beryllium 4	24 Mg magnesium 12		Sr strontium 38	137 <b>Ba</b> barium 56	[226] <b>Ra</b> radium 88
	_		7 Li lithium 3		39 K potassium 19	85 <b>Rb</b> rubidium 37	133 Cs caesium 55	[223] Fr francium 87

\* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.

The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.